

## Ch. 1 Notes (part 1) – THE CENTRAL SCIENCE

NOTE: Vocabulary terms are in **boldface and underlined**. Supporting details are in *italics*.

- I. Chemistry
  - A. Why study chemistry?  
CHEM IS LIFE! Chemistry is the study of everything around us.  
Everything in the natural world can be connected to chemistry.
  - B. What is **chemistry**? some definitions...
    - 1) *the study of matter and its changes*
    - 2) a science that deals with the composition, structure, and properties of substances and their transformations
    - 3) scientific study of matter, its properties, and interactions with other matter and energy
    - 4) study of **substances** – matter with a definite composition (elements, compounds, etc.)
  - C. Some *branches of chemistry*
    - 1) *Analytical* – chemical composition
    - 2) *Astrochemistry* – of the cosmos (planetary atmospheres, comets, etc.)
    - 3) *Biochemistry* – rxns. (reactions) of living organisms
    - 4) *Environmental* – environmental issues: pollution, water quality, etc.
      - a) ozone ( $O_3$ ) vs. oxygen gas ( $O_2$ )
      - b) chlorofluorocarbons (CFCs)
      - c) acid deposition ( $H_2SO_4$ ,  $HNO_3$ )
    - 5) *Food and Cooking* – reactions during cooking, ingredients, etc.
    - 6) *Geochemistry* – geological issues; petrology, etc.
    - 7) *Industrial chemistry* – study and implementation of chemical processes in manufacturing goods
    - 8) *Inorganic* – focuses on inorganic compounds (without carbon)
    - 9) *Nuclear* – focuses on rxns. changing of matter to energy
    - 10) *Organic* – focuses on carbon compounds
    - 11) *Physical* – physical characteristics and reaction mechanisms
    - 12) *Polymer chemistry* – study of chemical composition of plastics
    - 13) *Thermochemistry* – thermodynamics; heat changes
    - 14) *Theoretical chemistry* – “the examination of the structural and dynamic properties of molecules and molecular materials using the tools of quantum chemistry, equilibrium and nonequilibrium statistical mechanics and dynamics” (from Cornell)

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- II. Chemistry and Matter
    - A. Mass and weight
      - 1) **matter**—*a substance that takes up space and has mass*
      - 2) **mass**—*the amount of matter in an object*
      - 3) *properties of matter*—characteristics and behavior; can be physical or chemical
      - 4) **weight** – *Earth’s gravitational pull on an object*
    - B. Structure and observable characteristics
      - 1) *macroscopic view*—matter large enough to be seen
      - 2) *submicroscopic view*—dealing with atoms

- 3) scientific **model**—a visual, verbal, or math explanation of data; helps you understand the relationship between macroscopic and submicroscopic views
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### III. Scientific Methods

#### A. Definitions of **Science** (both content and process)

- 1) a process of using evidence to construct testable hypotheses
- 2) process of attaining knowledge about the natural world through study and practice
- 3) knowledge about or study of the natural world based on facts learned through experiments and observation. (Webster)
- 4) “knowledge covering general truths of the operation of general laws, especially as obtained and tested through scientific method and concerned with the physical world” (from New Collegiate)

#### B. Science as a systematic approach to the process of learning

##### 1) **scientific method**

- a) *systematic plan for testing ideas*
- b) *an organized way to solve problems*

##### 2) **observation**—*gathering and recording information*

- a) *direct observation—made with the senses* (sight, sound, smell, touch, hearing)  
“It is hot in here.”
- b) *indirect observation—made with measuring instruments* (thermometers, rulers, scales, clocks, etc.) “It is 83° in the room.”
- c) **qualitative data**—*verbal, not numerical, description* (rough, bright, red... )
- d) **quantitative data**—*description by numbers* (10 lbs., 98.6°, 5'4" tall...)

##### 3) **hypothesis**

- a) *educated guess; testable prediction/explanation*
- b) can be accepted or rejected, not proven
- c) many are made initially; most likely ones are chosen to pursue

##### 4) experimental and control setups

- a) **experiment**—*a controlled test of a hypothesis*
- b) **experimental group**—*the variable being tested is present in this group*
- c) **control group**—*the variable being tested is absent from this group*

##### 5) **variables**

- a) *anything affecting the outcome of the experiment*
- b) examples: temperature, air quality, amount of light, humidity
- c) *only one can be tested at a time for the experiment to be valid*
- d) **independent variable**
  - *changed by the experimenter*
  - abscissa: x axis
- e) **dependent variable**
  - *changes based on what the independent variable does*
  - ordinate: y axis

##### 6) **conclusion**—*a judgment based on information gathered during an experiment*

##### 7) What happens next? More research!

- a) review the existing literature
- b) experimental results are shared with other scientists
- c) repeat experiments to see if results are consistent

8) **theory**

- a) *repeatedly and thoroughly tested and supported explanation*
- b) *long description which tells why*
- c) *can never be proven*

The layman's definition of theory is incorrect! ("I have a theory why they aren't talking to me.")  
Theories are not guesses, nor are they wild ideas.  
Real theories have substantial scientific evidence behind them.

9) **scientific law**

- a) *concise statement which tells what*
- b) *relationship in nature that is supported by many experiments' results*
- c) *can be proven*

If a question asks for *lab design*, include the following:

- a. Your *hypothesis* and/or predictions/expected results
- b. The *independent variable* - what treatments will you apply
- c. The **dependent variable** - what will you measure
- d. The **variables to be controlled** (very important)
- e. The **organism/materials/apparatus to be used**
- f. Describe what you will **actually do**
- g. Describe how you will actually **take and record data**
- h. Describe how the data will be **graphed and analyzed**
- i. State how you will draw a **conclusion** (compare results to hypothesis and predictions)

Note: Your experimental design **needs to be at least theoretically possible** and it is very important that your conclusions/predictions be consistent with the principles involved and with the way you set up the experiment.

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IV. Scientific research

- A. **Pure research**—investigation for the sake of knowledge
- B. **Applied research**—investigation to solve a specific problem